

Nomenclature Practice

Name	Formula
1. lithium oxide	_____
2. cobalt (II) nitrite	_____
3. iron (III) sulfate	_____
4. dinitrogen dioxide	_____
5. ammonium phosphide	_____
6. potassium cyanide	_____
7. tin (IV) chromate	_____
8. cadmium arsenide	_____
9. mercury (II) perchlorate	_____
10. iodous acid	_____
11. nickel (II) peroxide	_____
12. strontium acetate	_____
13. phosphoric acid	_____
14. chromium (III) hydrogen carbonate	_____
15. hydrochloric acid	_____
16. magnesium hydroxide	_____
17. iodine trifluoride	_____
18. zinc phosphate	_____
19. fluoric acid	_____
20. silver permanganate	_____
21. trisulfur nonoxide	_____
22. rubidium nitride	_____
23. silicon dichloride	_____
24. lead (II) sulfite	_____
25. copper (I) dichromate	_____
26. cesium carbonate	_____
27. nitrous acid	_____
28. hydrogen bromide	_____
29. sodium hydride	_____
30. aluminum hydrogen phosphate	_____

Formula**Name**

1. HClO (aq)

2. BeOH

3. HF (aq)

4. H₂S (g)

5. Cr₂(SO₄)₃

6. SO₃

7. Ni₃(PO₄)₂

8. Ti(IO₃)₄

9. Hg₂(NO₂)₂

10. MnP

11. (NH₄)₂O₂

12. P₂O₅

13. H₂SO₃ (aq)

14. CaCrO₄

15. Pb(CO₃)₂

16. MnO₃

17. Ag₂S

18. Na₂HPO₄

19. CuSe

20. Al(C₂H₃O₂)₃

21. FCl₅

22. Fe₂O₃

23. XeF₄

24. HFO₄(aq)

25. CoN

26. Ba(CN)₂

27. MgI₂

28. NH₄OH

29. LiH

30. AsCl₃
