

Nomenclature Review

Please name the following compounds:

1. $\text{Fe}(\text{NO}_3)_2$ _____
2. MgCl_2 _____
3. H_2SO_3 (aq) _____
4. HgF_2 _____
5. PbS_2 _____
6. CO_3 _____
7. CaNO_2 _____
8. HI (aq) _____
9. AgCl _____
10. $\text{Cu}(\text{C}_2\text{H}_3\text{O}_2)_2$ _____
11. HCN (aq) _____
12. N_2O_5 _____
13. Sn_3N_4 _____
14. P_2S_4 _____
15. K_2CrO_4 _____
16. $(\text{NH}_4)_2\text{SO}_4$ _____
17. $\text{Be}(\text{OH})_2$ _____
18. MnO_2 _____
19. $\text{Al}_2\text{Cr}_2\text{O}_7$ _____
20. NaMnO_4 _____
21. HNO_2 (aq) _____
22. IBr _____
23. ZnCO_3 _____
24. Hg_2S _____
25. Li_3PO_4 _____

Please write the correct formula for the following compounds:

1. ammonium sulfite _____
2. hydrosulfuric acid _____
3. copper (I) nitride _____
4. lead (II) chloride _____
5. dinitrogen hexafluoride _____
6. carbonic acid _____
7. zinc hydroxide _____
8. lithium dichromate _____
9. magnesium phosphide _____
10. calcium acetate _____
11. iron (III) nitrite _____
12. aluminum sulfide _____
13. manganese (IV) chromate _____
14. cyanic acid _____
15. carbon monoxide _____
16. sulfur tetrachloride _____
17. mercury (I) nitrate _____
18. silver permanganate _____
19. tin (II) oxide _____
20. potassium phosphate _____
21. beryllium sulfate _____
22. hydrofluoric acid _____
23. cobalt (III) iodide _____
24. water _____
25. cadmium cyanide _____